Problem Set #17

Due: Thursday, 9 February 2012

- If we think of an electron as a particle, the function P(r) := 1 (2r² + 2r + 1)e^{-2r} is the cumulative distribution function of the distance r of the electron in a hydrogen atom from the center of the atom. (The distance is measured in Bohr radii; 1 Bohr radius = 5.29 × 10⁻¹¹ m. Niels Bohr (1885-1962) was a Danish physicist.) For example, P(1) = 1 5e⁻² ≈ 0.32 means that the electron is within 1 Bohr radius from the center of the atom 32% of the time.
 - (a) Find a formula for the density function of this distribution. Sketch the density function and the cumulative distribution function.
 - (b) Find the median distance and the mean distance. Near what value of *r* is an electron most likely to be found?
- 2. Suppose that a < b. The purpose of this problem is to show that if f is a quadratic polynomial, then we have

$$\int_{a}^{b} f(x) \, dx = \frac{b-a}{3} \left(\frac{f(a)}{2} + 2f\left(\frac{a+b}{2}\right) + \frac{f(b)}{2} \right) \, .$$

- (a) Show that this equation holds for $f_0(x) = 1$, $f_1(x) = x$ and $f_2(x) = x^2$.
- (b) Show that the equation holds for any quadratic polynomial $f(x) = Ax^2 + Bx + C$.
- **3.** Consider the following method for approximating $\int_{a}^{b} f(x) dx$. Partition the interval [a, b] into *n* equal subintervals. On each subinterval approximate the function *f* by a quadratic polynomial that agrees with *f* at both endpoints and at the midpoint of the subinterval.
 - (a) Explain why the integral of f on the subinterval $[x_{i-1}, x_i]$ is approximately equal to the expression

$$\frac{x_i - x_{i-1}}{3} \left[\frac{f(x_{i-1})}{2} + 2f\left(\frac{x_{i-1} + x_i}{2}\right) + \frac{f(x_i)}{2} \right]$$

(b) Show that if we add up these approximations, we get Simpson's rule:

$$\int_{a}^{b} f(x) \, dx \approx \frac{2}{3} \text{MID}(n) + \frac{1}{3} \text{TRAP}(n) \, .$$